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The compelling need to [Home](#) » [About us](#) » Technology reduce CO<sub>2</sub> emissions has led to many creative schemes to accomplish CO<sub>2</sub> emission reduction. For economic or social reasons, most are simply not feasible. However, EnPro's CO<sub>2</sub> emission reduction technology is a monumental breakthrough, elegant in its simplicity, effective in practice. Unlike any other solution proposed thus far, EnPro's process has some important benefits:

- Sequestration of CO<sub>2</sub> emissions by over 95%.
- Non-ammonia process for production of high quality 'green' soda ash (Na<sub>2</sub>CO<sub>3</sub>) – a stable industrial raw material of commercial value; low energy consumption – current estimates indicate energy consumption of less than half that of the current processes for production of soda ash.
- New low energy non-electrolysis process for production of caustic soda (NaOH).
- Technology is flexible enough for the plant to produce either soda ash or sodium hydroxide.

### The EnPro process at a glance

EnPro holds the exclusive global user rights to proprietary CO<sub>2</sub> emission reduction technology, which is based on chemical reactions used to consume CO<sub>2</sub> from exhaust gases. In general terms, the EnPro process removes CO<sub>2</sub> by chemically binding it to the salt in seawater. Valuable end-products such as sodium carbonate (soda ash) and/or sodium bicarbonate (baking soda) are extracted from the process. This is how it works:

- Salt water, together with salt, goes into the production of sodium hydroxide solution.
- CO<sub>2</sub> is sparged into the sodium hydroxide solution in a so-called Carbonation module where chemical reactions facilitate a binding of the CO<sub>2</sub> to the sodium in the solution.

- This results in sodium carbonate, which further processed into sodium carbonate (soda ash).
- The by-products are hydrochloric acid and hypochlorite.

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## About EnPro

EnPro AS is a Norwegian environmental technology company with a proven revolutionary new technology in producing stable minerals, by using CO<sub>2</sub> waste as a key raw material, with large CO<sub>2</sub> reduction impact.

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